

Horiba XGT9000: Micro X-ray fluorescence instrument (μ XRF)

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Photographs of Horiba XGT9000 instrument and its internal structure



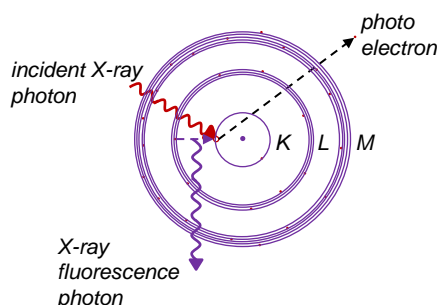
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Specifications of Horiba XGT9000

| Parameter | Value |
|---------------|--|
| X-ray source | Rh-K α : 20.2 keV |
| Detector | SDD: Resolution @MnK α ~ 130 eV |
| Tube voltage | 15 kV, 30 kV, 50kV |
| Tube current | 20 μ A – 1000 mA |
| Spot size | $\geq 10 \mu$ m |
| Analyzed area | $\leq (100 \times 100) \text{ mm}^2$ |
| Sample type | Solid (flat), powder, liquid |
| Atmosphere | Vacuum and air |

X-ray fluorescence (XRF)

When high-energy X-rays from a controlled source strike a sample, they can eject electrons from the inner shell of an atom. The atom stabilises by filling the gap with an electron from a higher energy level, emitting a characteristic X-ray photon known as X-ray fluorescence (XRF). Measuring the energy of these photons reveals the elemental composition of the sample. Counting photons of a certain energy (measuring intensity) leads to quantification.

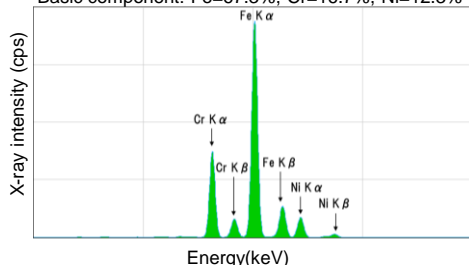


Excitation and relaxation by XRF

Micro X-ray fluorescence (μ XRF)

The μ XRF instrumentation consists of X-ray focusing optics to achieve a fine beam size and a movable sample stage, which work together to produce laterally resolved elemental maps of (flat) samples.

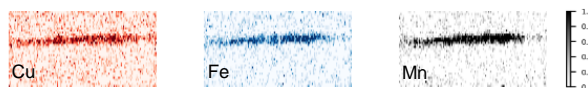
Sample: stainless steel (SUS304)
Basic component: Fe=67.3%, Cr=16.7%, Ni=12.3%



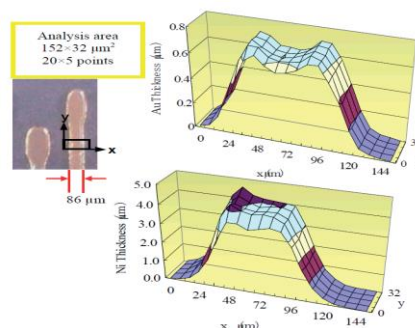
XRF spectrum measured on stainless steel (©Horiba Ltd.)

μ XRF in battery research

μ XRF in battery research can be used to study elemental composition and distribution of elements with atomic numbers $z \geq 11$ with lateral resolution of approx. 10μ m. Electrode electrodes, electrolyte and separators can be studied providing an element sensitivity down to sub-ppm range. This enables the investigation of homogeneity, impurities, dissolution and deposition of (active) elements. Furthermore, (metal) film thickness on a substrates can be studied.



Distribution of Cu, Fe, and Mn (active elements) along the thickness of electrode $\text{Na}_{0.9}\text{Mn}_{0.48}\text{Fe}_{0.3}\text{Cu}_{0.22}\text{O}_2$ (measured on Horiba XGT9000).



Variations of Au and Ni layer thicknesses deposited on Cu substrate, acquired across the region shown in the optical image (© Horiba Ltd.)

Reference

1. www.horiba.com/int/scientific/
2. Sadique *et al.* MRS Advances (2022)7:361-365.
3. Evertz *et al.* Journal of Power Sources (2016)329:364-371